

THE CHEMISTRY OF GEMSTONE COLOURS

Gemstone colours stem from their chemical structures, which absorb different wavelengths of light. Their hardness is measured on the Mohs hardness scale (1-10).

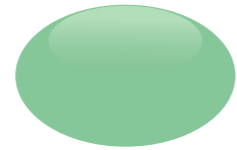


PEARL

Formula: CaCO_3

Mohs hardness: 2.5–4.5

Produced in soft tissue of shelled molluscs. The thinner the layers of the pearl, the finer the lustre.

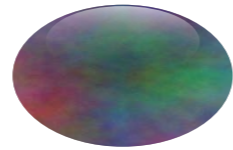


TURQUOISE

Formula: $\text{Al}_6(\text{PO}_4)_4(\text{OH})_8 \cdot 4\text{H}_2\text{O}$

Mohs hardness: 5.0–6.0

Colour caused by the presence of copper ions coordinated to the hydroxide ions and water.

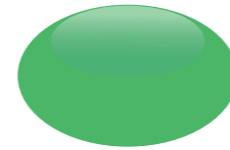


OPAL

Formula: $\text{SiO}_2 \cdot n\text{H}_2\text{O}$

Mohs hardness: 5.5–6.0

'Play of colours' caused by interference and diffraction of light passing through structure.

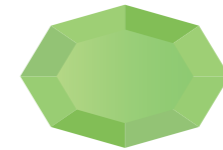


JADE

Formula: $\text{NaAlSi}_2\text{O}_6$

Mohs hardness: 6.5–7.0

Colour from chromium and iron impurities. The mineral nephrite is also referred to as jade.

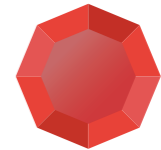


PERIDOT

Formula: Mg_2SiO_4

Mohs hardness: 6.5–7.0

Colour caused by iron 2+ ions replacing magnesium ions in some locations in the structure.

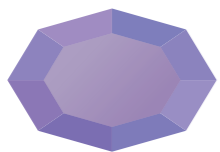


GARNET

Formula: $\text{Mg}_3\text{Al}_2(\text{SiO}_4)_3$

Mohs hardness: 6.5–7.5

Colour caused by iron 2+ ions replacing magnesium ions in some locations in the structure.

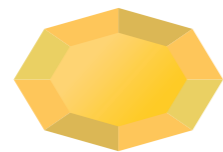


AMETHYST

Formula: SiO_2

Mohs hardness: 7.0

Colour caused by irradiation of iron 3+ ions in place of silicon in some locations in the structure.



CITRINE

Formula: SiO_2

Mohs hardness: 7.0

The yellow colour of citrine is due to the presence of either aluminium or iron impurities.

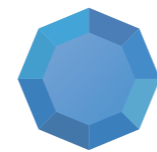


TOURMALINE

Formula: $\text{Na}_3\text{Li}_3\text{Al}_6(\text{BO}_3)_3(\text{SiO}_3)_6\text{F}_4$

Mohs hardness: 7.0–7.5

Colour due to manganese ions replacing lithium and aluminium ions in some sites.



ZIRCON

Formula: ZrSiO_4

Mohs hardness: 7.5

Many colours depending on impurities. Colourless forms are popular diamond substitutes.



AQUAMARINE

Formula: $\text{Be}_3\text{Al}_2(\text{SiO}_3)_6$

Mohs hardness: 7.5–8.0

Colour caused by iron 2+/3+ ions replacing aluminium ions in some locations in the structure.



EMERALD

Formula: $\text{Be}_3\text{Al}_2(\text{SiO}_3)_6$

Mohs hardness: 7.5–8.0

Colour caused by chromium ions replacing aluminium in some locations in the structure.



SPINEL

Formula: MgAl_2O_4

Mohs hardness: 7.5–8.0

A variety of colours are possible, caused by impurities such as iron, chromium and nickel.



TOPAZ

Formula: $\text{Al}_2\text{SiO}_4(\text{F},\text{OH})_2$

Mohs hardness: 8.0

Pure topaz is colourless; blue & brown varieties are caused by atomic level imperfections.



ALEXANDRITE

Formula: Al_2BeO_4

Mohs hardness: 8.5

Colour caused by chromium ions replacing aluminium in some sites. Colour varies in different light.



RUBY

Formula: Al_2O_3

Mohs hardness: 9.0

Colour caused by chromium ions replacing aluminium ions in some locations in the structure.



SAPPHIRE

Formula: Al_2O_3

Mohs hardness: 9.0

Colour caused by titanium and iron ions replacing aluminium ions in some locations in the structure.



DIAMOND

Formula: C_n

Mohs hardness: 10

Colourless; can be faintly coloured by the trapping of nitrogen or boron atoms in the crystal.

