THE CHEMISTRY OF SUNSCREEN

Summer sun brings with it the risk of sunburn, so we'll all be slapping on the sunscreen to guard against it. But what are the chemicals that keep you from turning as red as a lobster? This graphic looks at them and how they work.



TYPES OF UV RADIATION

wavelength 320-400nm

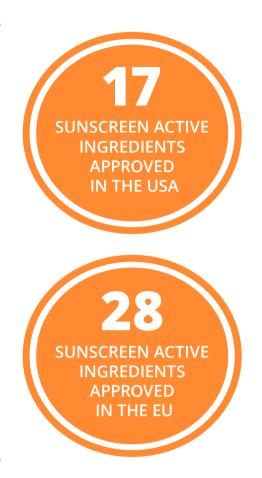
Accounts for 95% of solar UV radiation reaching Earth's surface. Penetrates deepest into skin, and contributes to skin cancer via indirect DNA damage.

UVB 290-320nm

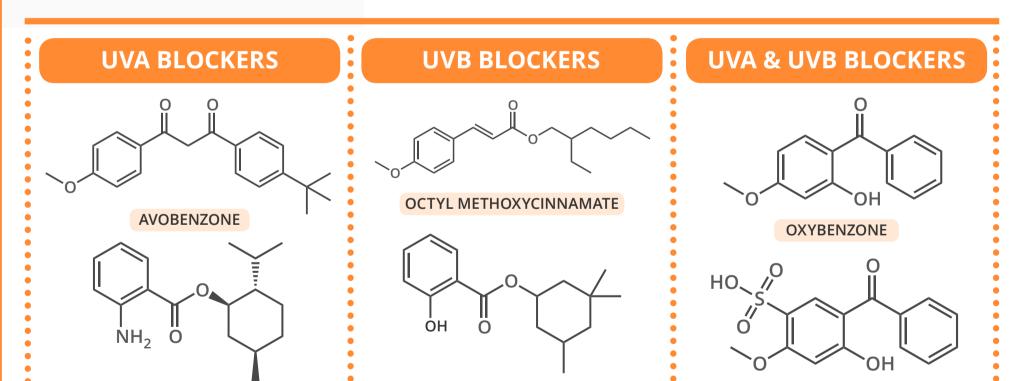
Accounts for 5% of solar UV radiation reaching Earth's surface. Causes direct DNA damage, and is one of the main contributors to skin cancer.

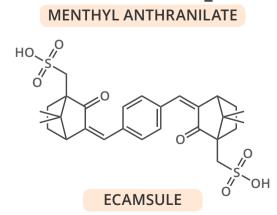
UVC 290-100nm

Filtered out by ozone in the Earth's atmosphere, and as a result does not reach the surface of the Earth, and doesn't cause skin damage.



Inorganic chemicals in sunscreen, such as zinc oxide and titanium oxide, both absorb and scatter UV light. Organic chemicals are also used – the chemical bonds in these absorb UV radiation, with the chemical structure affecting whether they absorb UVA, UVB, or both. Several different chemicals are used in sunscreen to ensure full protection.





HOMOSAL	ATE
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OTHER UVB BLOCKERS
(Italicised = not approved in USA)AOctylocrylene

Ensulizole

Octyl triazone

Enzacamene

- PABA Padimate O
- Cinoxate
- Octyl salicylate
- Trolamine salicylate Amiloxate

OTHER UVA & UVB BLOCKERS
(Italicised = not approved in USA)DioxybenzoneNeo Heliopan APMexoryl XLUvinul A PlusTinosorb SUVAsorb HEBTinosorb M

All currently approved in EU, Canada & Australia

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