Today in Chemistry History @

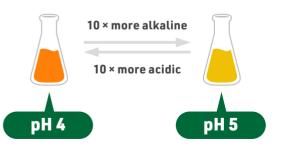
9th January – Søren Sørensen's birthday (1868) and the pH scale



The pH scale and what it means



The pH scale typically runs from 0 to 14 and is a way of measuring how acidic or alkaline a solution is. It's a logarithmic scale, so a change of one represents a ten times increase or decrease in acidity or alkalinity. A neutral solution has an equal concentration of hydrogen and hydroxide ions; this is usually at pH 7, but can vary slightly depending on temperature.



© (†) 😒 =

COMPO