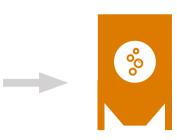
The Chemistry of Rum





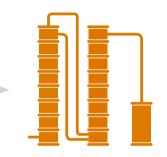
RAW MATERIALS

Molasses, sugar cane juice, or cane syrup



FERMENTATION

Usually at $30-33^{\circ}$ C and pH 5.5-5.8 for 1-3 days



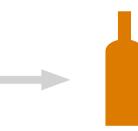
DISTILLATION

Using either continuous or pot-still distillation



AGEING

Often carried out in charred oak barrels



MIXING

Different distillates blended for consistency

Esters



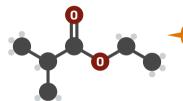
Ethyl propanoate

Caramel/fruity aroma

Ethyl acetate

Pear drops aroma

The levels of short-chain carboxylic acids are higher in rum than in other spirits like whiskey, which may explain its higher ester content when compared to other alcohols.

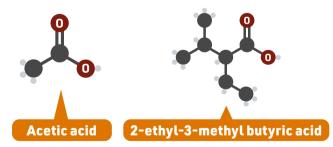


Ethyl isobutyrate

Butterscotch

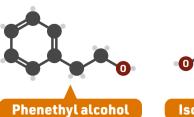
Esters are important contributors to the aroma of rum, and are responsible for fruity notes. Ethyl acetate is typically the most abundant, though a large number of other esters are also present.

Acids and alcohols



Acids in rum are important for production of esters, but also contribute to flavour.

Acetic acid is the main volatile acid in rum, whereas 2-ethyl-3-methyl butyric acid is characteristic of rums. Strong smelling higher alcohols are also important odorants.



Isoamyl alcohol

Floral aroma Malty aroma



KEY

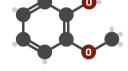
Carbon



Hydrogen

Other compounds





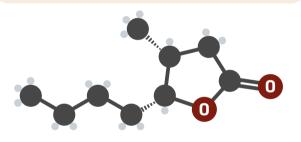
Vanillin

Sweet, vanilla aroma



Smoky aroma

A number of compounds originate from the barrels that the rum is aged in. These include phenolic compounds, and also oak lactones. These are found in lower quantities than in whiskey, as rum is not aged in barrels for as long.

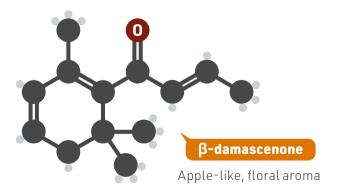


(Z)-oak lactone

Woody aroma

Dark vs. white rum

To remove colouration, white rums are usually filtered through charcoal. As well as removing colour-causing compounds, this also removes some aroma and flavour molecules.



β-damascenone is a potent odorant in dark rum due to its low odour threshold. However, it is much less apparent in white rum, due to loss during filtration.

